1. Let $P = P(t)$ be the population at time t, and P_0 the initial population. From $dP/dt = kP$ we obtain $P = P_0e^{kt}$. Using $P(5) = 2P_0$ we find $k = \frac{1}{5}\ln 2$ and $P = P_0e^{(\ln 2)t/5}$. Setting $P(t) = 3P_0$ we have $3 = e^{(\ln 2)t/5}$, so

$$\ln 3 = \frac{(\ln 2)t}{5} \qquad \text{and} \qquad t = \frac{5\ln 3}{\ln 2} \approx 7.9 \text{ years.}$$

Setting $P(t) = 4P_0$ we have $4 = e^{(\ln 2)t/5}$, so

$$\ln 4 = \frac{(\ln 2)t}{5} \qquad \text{and} \qquad t \approx 10 \text{ years.}$$

3. Let $P = P(t)$ be the population at time t. Then $dP/dt = kP$ and $P = ce^{kt}$. From $P(0) = c = 500$ we see that $P = 500e^{kt}$. Since 15% of 500 is 75, we have $P(10) = 500e^{10k} = 575$. Solving for k, we get $k = \frac{1}{10}\ln\frac{575}{500} = \frac{1}{10}\ln 1.15$. When $t = 30$,

$$P(30) = 500e^{(1/10)(\ln 1.15)30} = 500e^{3\ln 1.15} = 760 \text{ years}$$

and

$$P'(30) = kP(30) = \frac{1}{10}(\ln 1.15)760 = 10.62 \text{ persons/year.}$$

5. Let $A = A(t)$ be the amount of lead present at time t. From $dA/dt = kA$ and $A(0) = 1$ we obtain $A = e^{kt}$. Using $A(3.3) = 1/2$ we find $k = \frac{1}{3.3}\ln(1/2)$. When 90% of the lead has decayed, 0.1 grams will remain. Setting $A(t) = 0.1$ we have $e^{t(1/3.3)\ln(1/2)} = 0.1$, so

$$\frac{t}{3.3}\ln\frac{1}{2} = \ln 0.1 \qquad \text{and} \qquad t = \frac{3.3\ln 0.1}{\ln(1/2)} \approx 10.96 \text{ hours.}$$

11. Assume that $A = A_0e^{kt}$ and $k = -0.00012378$. If $A(t) = 0.145A_0$ then $t \approx$15,600 years.

13. Assume that $dT/dt = k(T - 10)$ so that $T = 10 + ce^{kt}$. If $T(0) = 70°$ and $T(1/2) = 50°$ then $c = 60$ and $k = 2\ln(2/3)$ so that $T(1) = 36.67°$. If $T(t) = 15°$ then $t = 3.06$ minutes.

15. We use the fact that the boiling temperature for water is 100° C. Now assume that $dT/dt = k(T - 100)$ so that $T = 100 + ce^{kt}$. If $T(0) = 20°$ and $T(1) = 22°$, then $c = -80$ and $k = \ln(39/40) \approx -0.0253$. Then $T(t) = 100 - 80e^{-0.0253t}$, and when $T = 90$, $t = 82.1$ seconds. If $T(t) = 98°$ then $t = 145.7$ seconds.

19. Identifying $T_m = 70$, the differential equation is $dT/dt = k(T - 70)$. Assuming $T(0) = 98.6$ and separating variables we find $T(t) = 70 + 28.9e^{kt}$. If $t_1 > 0$ is the time of discovery of the body, then

$$T(t_1) = 70 + 28.6e^{kt_1} = 85 \quad \text{and} \quad T(t_1 + 1) = 70 + 28.6e^{k(t_1+1)} = 80.$$

Therefore $e^{kt_1} = 15/28.6$ and $e^{k(t_1+1)} = 10/28.6$. This implies

$$e^k = \frac{10}{28.6}\, e^{-kt_1} = \frac{10}{28.6} \cdot \frac{28.6}{15} = \frac{2}{3}\,,$$

so $k = \ln \frac{2}{3} \approx -0.405465108$. Therefore

$$t_1 = \frac{1}{k} \ln \frac{15}{28.6} \approx 1.5916 \approx 1.6.$$

Death took place about 1.6 hours prior to the discovery of the body.